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Chapter 5 Vocab Modern Chemistry (Holt McDougal) STUDY. PLAY. Periodic Law. The physical and chemical properties of the elements are periodic functions of their atomic numbers. Periodic Table. An arrangement of the elements in order of their atomic numbers so that elements with similar properties fall in the same column, or group.

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

[5 The Periodic Law](#)

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Taken from the book Modern Chemistry by Holt, Rinehart, and Winston on Chapters 4 and 5, which deals with electrons and the periodic table. Includes the chapter vocabulary and a few other useful things. Terms : Hide Images A form of energy that exhibits wavelike behavior as it travels through space (3.00x10⁸ m/s)

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Atomic Structure - Ch. 3. Atomic Timeline Song I. Structure of the Atom II. Masses of Atoms. Matter - Ch. 1. I. States of Matter II. Classification of Matter III. Properties & Changes in Matter. Measurement - Ch. 2. I. Using Measurements II. Units of Measurement III. Unit Conversions

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Holt Modern Chemistry (0th Edition) Edit edition. Problem 51CR from Chapter 13: Find out how much salt a large northern city, such as New Yo... Get solutions

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